

## **Chemistry Content Standards:**

\_\_\_\_\_

The conservation of atoms in chemical reactions leads to the principle of conservation of matter and the ability to calculate the mass of products and reactants.

<u>Standard 3e</u>: Students know how to calculate the masses of reactants and products in a chemical reaction from the mass of one of the reactants or products and the relevant atomic masses.

12.2.1 Construct mole ratios from balanced chemical equations and apply these ratios in stoichiometric calculations

12.2.2 Calculate stoichiometric quantities from balanced chemical equations

## Stamp:

	Moles and Stoichiometry A Page 366, 5-8
•	WB 12.1
Lecture: Classwork :	Stoichiometry Lecture Part Two
Classwork :	Moles and Stoich Test B
Home Work:	WB 12.2 Page 366, 21-22