

VOCABULARY

Chemistry- Mr. Hunter

Week 6

Chapter 6

1. **Periodic Law:** When the elements are arranged in order of increasing atomic number.
2. **Metals:** one of a class of elements that are good conductors of heat and electric current.
3. **Nonmetals:** an element that tends to be a poor conductor of heat and electric currents.
4. **Metalloid:** an element that tends to have properties that are similar to those of metals and nonmetals.
5. **Alkali metals:** any metal in Group 1A of the periodic table.
6. **Alkaline Earth Metals:** any metal in Group 2A of the periodic table.
7. **Halogens:** a nonmetal in Group 7A of the periodic table.
8. **Noble Gases:** an element in 8A of the periodic table.
9. **Ions:** an atom or group of atoms that have a positive and negative charge.
10. **Cation:** any atom or group of atoms with a positive charge.
11. **Anion:** any atom or a group of atoms with a negative charge.
12. **Electronegativity:** the ability of an atom to attract electrons when the atom is in a compound.