

Chemistry Content Standards:

Energy is exchanged or transformed in all chemical reactions and physical changes of matter

Standard 7a: Students know how to describe temperature and heat flow in terms of the motion of molecules

Standard 7b: Students know chemical processes can either release exothermic or absorb endothermic thermal energy

Standard 7c: Students know energy is released when a material condenses or freezes and is absorbed when a material evaporates or melts.

Standard 7d: Students know how to solve problems involving heat flow and temperature changes, using known values of specific heat and latent heat of phase change.

- 17.2.1: Describe how calorimeters are used to measure heat flow
- 17.2.2: Construct thermochemical equations
- 17.2.3 Solve for enthalpy changes in chemical reactions by using heats of reaction.

Stamp:

Page 510, 5-11 WB 17.1 Practice Problems 17.1 SR 17.1

Lecture: Changes	Thermodynamics 17.2 Measuring and Expressing Enthalpy Thermochemical Calculations PPT Practice Problems 17.2
Class work:	Students should complete the following assignments for direct instruction, "I do" with student assessmentand individual/group learning
	 Calculations Involving Specific Heat WB 17.2

Homework: None