## LP Wednesday, October 2<sup>nd</sup> and Thursday, October 3<sup>rd</sup>, 2019 Unit 3.2



## CALIFORNIA CONTENT STANDARDS:

1g: Students know that spectral lines are the result of transitions of electrons between energy levels and that these lines correspond to photons with a frequency related to the energy spacing between levels by using Planck's relationship, (E=hv)

## CALIFORNIA CONTENT STANDARDS:

- 5.3.1: Describe the relationship between the wavelength and frequency of light.
- 5.3.2: Identify the source of atomic emission spectra
- 5.3.3: Explain how the frequencies of emitted light are related to changes in electron energies.

Stamp WB:	Have students pass work to the north of lab, TAs will stamp work
	Electron Configuration WB Page 154-155
Class	Hand out Practice Problems 5.1 (20 Min)
Review:	Energy Levels of Hydrogen Students take out piece of paper. Complete the three practice problems in 5.1 Go over answers once class is finished Collect handouts after work is completed
Video:	Bill Nye: Light and Color (30 Min) Hand out Video Questions Have students answer questions while watching video Collect Handouts
Video:	Bill Nye: Light Optics Hand out Video Questions Have students answer questions while watching video Collect handouts
Home Work:	None